Experimental Evidence for the Existence of $SiCH_x$ (x = 1-3) Molecules in the Gas Phase[†]

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Abstract: The neutral silicon carbon hydrides SiCH_x (x = 1-3) are viable molecules in the diluted gas phase as evidenced by neutralization-reionization mass spectrometry experiments. Thus, previous suggestions concerning conceivable mechanisms for the generation of these molecules in interstellar space are strongly supported.

The chemistry of organic silicon compounds is one of the fastest growing areas in organic chemistry.¹ Among the several exciting new developments the synthesis of molecules containing multiple bonds to silicon² is particularly noteworthy as is the fruitful interplay between experiment and theory, with the latter often preceding and sometimes even guiding experimental work.³ The chemistry of silicon compounds is also viewed as a fundamental problem in astrophysics and astrochemistry.⁴ Here, the understanding of chemical pathways from atomic silicon cation $(Si^+(^2P))$ to neutral SiCH_x species is of special interest and importance.⁵ Several processes have been suggested for the formation of, for example, SiCH[•], SiCH₂, and SiCH₃[•], respectively. One involves the transfer of a "carbanionic fragment"6 to Si*+, a reaction which is often driven by the formation of stabilized cations (eq 1).

$$\mathrm{Si}^{*+} + (\mathrm{CH}_3)_3 \mathrm{N} \rightarrow \mathrm{Si}\mathrm{CH}_3^* + \mathrm{CH}_2 \mathrm{NH}\mathrm{CH}_3^+ \qquad (1)$$

A different route may involve dissociative neutralization of $SiCH_x^+$ species by either proton detachment (eq 2) or electron capture (eq 3).

$$\operatorname{SiCH}_{x^{+}} - \underbrace{\operatorname{SiCH}_{(x-1)} + \operatorname{H}^{+}}_{0^{-}} \operatorname{SiCH}_{x^{+}} - \operatorname{SiCH}_{(x-1)} + \operatorname{H}^{\bullet} (3)$$

However, the evidence for the actual formation of neutral SiCH_x is in most cases at best circumstantial.⁷ The failure of directly probing the existence of stable $SiCH_x$ species does, of course, not necessarily imply an intrinsic instability of these molecules. Rather, it may reflect a high propensity to undergo facile intermolecular reactions which prevent a successful detection of these electronically unsaturated fragments.

In fact, high-level ab initio MO calculations⁸ leave no doubt that, for example, SiCH* is ca. 60 kcal/mol more stable than its neutral isomer HSiC*.8b Similarly, on the theoretically well-studied singlet and triplet potential energy surfaces of [Si,C,H₂] several minima were located.^{8a,c} For example, the global minimum is predicted to correspond to 1-silavinylidene, SiCH₂; this species is 50 kcal/mol more stable than the trans-bent form of silaacetylene, HCSiH, which at the highest level of theory may not even exist as a minimum. The third isomer H₂SiC is calculated to be even higher in energy (90 kcal/mol relative to SiCH₂). Thus under appropriate conditions, e.g. in the diluted gas phase, $SiCH_x$ species are expected to be viable molecules and, as stated by Schaefer, ^{3a} "experiments probing the theoretical predictions would be particularly welcome". This is precisely what the present paper is aiming at.

By using the relatively recent technique of neutralizationreionization mass spectrometry (NRMS)⁹ we will demonstrate that SiCH_x (x = 1-3) are indeed viable species residing in potential wells deep enough to verify their existence. We were encouraged to conduct these gas-phase experiments by the recent observations that. by using NRMS, several otherwise elusive organometallic fragments were generated and characterized as genuine, chemically

bound molecules with lifetimes $t > 1 \ \mu s$. Typical examples include the following species: FeCH_x (x = 0-3), NiL_x complexes (x =1, 2; L = CO, D₂O, ND₃),¹¹ "end-on" Cu⁽⁰⁾ complexes of HCN and HNC,¹² the half-sandwiches MC_5H_5 (M = Fe, Co, Ni),¹³ the long sought after metal acetylides MCCH (M = Fe, Co, Ni),¹⁴ and the theoretically predicted organoaluminium fragments AlCH₂, Al(CH₃)_x (x = 1, 2), and AlC₂H₄, respectively.¹⁵

These and many other seemingly "unstable" molecules were accessible by first neutralizing a beam of ions in a collision experiment followed by reionization of the fast moving beam through

(1) For an excellent recent overview, see: Patai, S.; Rappoport, Z. The chemistry of functional groups: The chemistry of silicon compounds, Parts 1 and 2; Wiley Interscience: New York, 1989.

(2) For leading references, see: (a) Raabe, G., Michl, J. Chem. Rev. 1985,

(3) (a) Schaefer, H. F., III Acc. Chem. Res. 1982, 15, 283.
(b) Raabe, G.; Michl, J., in ref 1, Chapter 17.
(c) (a) Schaefer, H. F., III Acc. Chem. Res. 1982, 15, 283.
(b) Apeloig, Y., in ref 1, Chapter 2.
(c) Bock, H.; Solouki, B. ref 1, Chapter 9.
(d) Bock, H. Angew. Chem., Int. Ed. Engl. 1989, 28, 1627.

(4) (a) Bohme, D. K.; Wlodek, S.; Fox, A. In Rate Coefficients in As-trochemistry; Millar, T. J., Williams, D. A., Eds.; Kluwer Academic Publisher: Amsterdam, 1988; p 193. (b) Wlodek, S.; Bohme, D. K. J. Am. Chem. Soc. 1988, 110, 2396. (c) Bohme, D. K. Int. J. Mass Spectrom. Ion Processes 1990, 100, 719. (d) Herbst, E.; Millar, T. J.; Wlodek, S.; Bohme, D. K. Astron. Astrophys. 1989, 222, 205.

(5) (a) Turner, J. L.; Dalgarno, A. Astrophys. J. 1977, 213, 386. (b)
Millar, T. J. Astrophys. Space Sci. 1980, 72, 509. (c) Clegg, R. E. S.; v. ljzendoorn, L. J.; Allamandola, L. J. Mon. Not. R. astr. Soc. 1983, 203, 125. (d) Cernicharo, J.; Gottlieb, C. A.; Guélin, M.; Thaddeus, P.; Vrtilek, J. M. Astrophys. J. 1989, 341, L25. (e) Lim, K. P.; Lampe, F. W. Int. J. Mass Spectrom. Ion Processes 1990, 101, 245.
(f) A convine methonide transfer to meta-lien control has been computed.

(6) A genuine methanide transfer to metal-ion centers has been reported recently: (a) Georgiadis, R.; Fisher, E. R.; Armentrout, P. B. J. Am. Chem. Soc. 1989, 111, 4251.
(b) Weil, D. A.; Wilkins, C. L. J. Am. Chem. Soc. 1985, 107, 7316.
(c) Karraβ, S.; Schwarz, H. Int. J. Mass Spectrom. Ion Processes 1990, 98, R1.

(7) For an elegant, though indirect, "proof" that SiCH₂ is a stable gasphase molecule, see: Damrauer, R.; DePuy, C. H.; Barlouy, S. E.; Gronert, S. J. Am. Chem. Soc. 1988, 110, 2005. In this article, the authors report the gas-phase basicity of the silaacetylide anion, HCSi-; their data, in conjunction with ab initio MO calculations, demonstrate that protonation of HCSi⁻ should give rise to H₂CSi rather than the isomeric silaacetylene HCSiH.

(8) For a detailed discussion and many leading references, see: (a) Reference 3b. (b) Bruna, P. S.; Hirsh, G.; Buenker, R. J.; Peyerimhoff, S. D. In Molecular Ions; Berkowitz, J., Groeneveld, K., Eds.; Plenum: New York, 1983; p 309. (c) Luke, B. T.; Pople, J. A.; Krogh-Jespersen, M.; Apeloig, Y.; Karni, M.; Chandrasekhar, J.; Schleyer, P. v. R. J. Am. Chem. Soc. 1986, 108, 270. (d) Hoffmann, M. R.; Yoshioka, Y.; Schaefer, H. F., 111 J. Am.

108, 270. (d) Hoffmann, M. R.; Yoshioka, Y.; Schaefer, H. F., III J. Am. Chem. Soc. 1983, 105, 1084.
(9) Reviews: (a) Terlouw, J. K.; Burgers, P. C.; v. Baar, B. L. M.; Weiske, T.; Schwarz, H. Chimia 1986, 40, 357. (b) Wesdemiotis, C.; McLafferty, F. W. Chem. Rev. 1987, 87, 483. (c) Terlouw, J. K.; Schwarz, H. Angew. Chem., Int. Ed. Engl. 1987, 26, 805. (d) Schwarz, H. Pure Appl. Chem. 1989, 61, 685. (e) Holmes, J. L. Adv. Mass Spectrom. 1989, 11, 53. (f) Terlouw, J. K. Ibid. 1989, 11, 984. (g) Holmes, J. L. Mass Spectrom. Rev. 1989, 8, 513.
(h) McLafferty, F. W. Science 1990, 247, 925.
(10) Lebrilla C. B.; Drewello T.; Schwarz, H. Organometallics 1987, 6

(10) Lebrilla, C. B.; Drewello, T.; Schwarz, H. Organometallics 1987, 6, 2268.

(11) Hudgins, D. M.; Porter, R. F. Rap. Commun. Mass Spectrom. 1988, 2, 197.

(12) Eller, K.; Sülzle, D.; Schwarz, H. Chem. Phys. Lett. 1989, 154, 443. (13) Drewello, T.; Schwarz, H. Int. J. Mass Spectrom. Ion Processes 1989, 93, 177

(14) Drewello, T.; Schwarz, H. Chem. Phys. Lett. 1990, 171, 5. (15) Srinivas, R.; Sülzle, D.; Schwarz, H. J. Am. Chem. Soc., in press, and references therein.

subjecting it to a second collision experiment. As a primary ion beam we used $SiCH_x^+$ (x = 1-3) (see below). According to both theory^{3b,8b,16} and experiments¹⁷ the $SiCH_x^+$ cations are significantly more stable than their isomers $HSiCH_{x-1}^+$. For example, the energy difference between $SiCH^+$ and $HSiC^+$ excedes 70 kcal/ mol, and $SiCH_3^+$ is found to be by >57 kcal/mol more stable than $HSiCH_2^+$ (if the latter species corresponds at all to a true minimum). For the $[Si,C,H_2]^+$ system the lowest energy isomer is, according to Armentrout et al.,^{17b} almost certainly $SiCH_2^+$.

Experimental Section

The mass spectrometric experiments were performed using our modified ZAB-HF machine which is of BEBE configuration (B stands for magnetic and E for electric sector).¹⁸ SiCH_x⁺ ions were generated by magnetic and E for electric sector). A SiCH₂ Clemploying the following conditions: ion source temperature, 200 °C; trap current, 100 μ A; repeller voltage, 20 V; accelerating voltage, 8 kV; mass resolution $m/\Delta m$, 1500. Collision induced dissociations (CIDs),¹⁹ used to characterize the primary ions $SiCH_x^+$, were brought about by mass-selecting a beam of these ions by means of B(1)E(1) and colliding it with helium (80%) transmission, T; this transmittance corresponds, on the average, to 1.1-1.2 collisions²⁰). Ionic dissociation products were recorded by scanning B(2). In the NRMS experiment, a beam of BE-mass selected $SiCH_x^+$ ions (having 8-keV translational energy) was neutralized in the first cell of a differentially pumped tandem collision cell by colliding it with xenon (75% T). Unreacted ions were deflected away from the beam of neutral species SiCH_x by putting a voltage on a deflector electrode; subsequent reionization of SiCH, in a second collision cell took place by collision with oxygen (80% T). The mass spectrum of the resulting ions SiCH_x⁺ was recorded by scanning B(2). The minimal lifetime t (identical with the transit time from collision cell I to cell II) in the present experiment is ca. $0.5-1 \mu s$. The availability of a multisector machine and the fact that collision chambers are present in the field-free regions between B(1)/E(1) and E(1)/B(2) enabled us to perform a further experiment of the following kind: SiCH_x⁺ ions are mass-selected by B(1), neutralized with Xe (80% T), and reionized with oxygen (80% T) using the tandem collision cell located between B(1) and E(1). The reionized SiCH_x⁺ are subjected to a kinetic energy analysis by means of E(1), and the interference-free beam of $SiCH_x^+$ species of 8-keV kinetic energy was collisionally activated with helium (80% T) in the collision cell present in the field-free region E(1)/B(2). The CA mass spectra, which basically contain no contribution from reionized neutral fragments, were recorded by scanning B(2); these CA mass spectra of reionized species together with the presence of recovery signals in the NR mass spectra of SiCH.+ permit a direct identification of neutral SiCH_x. Signal-averaging techniques were used to improve the S/N ratio. Between 20 and 50 scans were accumulated by on-line processing the data with the VG 11/250 or the AMD-Intectra systems.

Results and Discussion

The SiCH⁺ species $(m/z \ 41)$ is easily accessible by 100-eV electron impact ionization of H₃SiCH₂Cl. Its CA mass spectrum contains signals at $m/z \ 40$ (SiC⁺⁺; 48% total fragment ion current), $m/z \ 29$ (SiH⁺, 5%), $m/z \ 28$ (Si⁺⁺, 47%), $m/z \ 13$ (CH⁺, <1%). In view of the above-mentioned high-energy difference between SiCH⁺ (1⁺) and HSiC⁺ (1a⁺) we presume that the CA spectrum of $m/z \ 41$ is in keeping with a connectivity SiCH⁺. The relatively small abundance of the $m/z \ 29$ ion is perhaps due to a partial collision induced isomerization SiCH⁺ \rightarrow HSiC⁺ \rightarrow $m/z \ 29$. In addition, as the mass resolution $m/\Delta m = 1500$ (all slits open to provide a good sensitivity for the multi-collision experiments) is not high enough to completely separate the doublet of $m/z \ 41$ ([H,²⁸Si,C]⁺ versus [²⁹Si,C]⁺⁺), some fraction of $m/z \ 29$, in all spectra discussed in this study, may well correspond to ²⁹Si⁺⁺.

(16) (a) Hopkinson, A. C.; Lien, M. H. J. Chem. Soc., Chem. Commun. 1980, 107. (b) Raghavachari, K. J. Phys. Chem. 1988, 92, 6284 and references therein.

(17) (a) Mandich, M. L.; Reents, W. D.; Bondebey, V. E. J. Phys. Chem. 1986, 90, 2315. (b) Boo, B. H.; Elkind, J. L.; Armentrout, P. B. J. Am. Chem. Soc. 1990, 112, 2083 and references therein.

(18) For a detailed description of the basic 3-sector machine, see: (a) Weiske, T. Ph.D. Thesis, Technische Universität Berlin D 83, 1985. (b) Terlouw, J. K.; Weiske, T.; Schwarz, H.; Holmes, J. L. Org. Mass Spectrom. 1986, 21, 665.

(19) (a) Levsen, K.; Schwarz, H. Angew. Chem., Int. Ed. Engl. 1976, 15, 509.
(b) Cooks, R. G., Ed. Collision Spectroscopy; Plenum: New York, 1978.
(c) Levsen, K.; Schwarz, H. Mass Spectrom. Rev. 1985, 3, 77.

(c) Levsen, K.; Schwarz, H. Mass Spectrom. Rev. 1985, 3, 77. (20) Holmes, J. L. Org. Mass Spectrom. 1985, 20, 169.



m/z-

Figure 1. NR mass spectrum of SiCH⁺, m/z 41 (xenon, 80% T//oxygen 80% T).





Figure 2. CA mass spectrum of reionized SiCH $^{\bullet}$ (xenon, 80% T//oxygen 80% T//helium 80% T).



Figure 3. NR mass spectrum of SiCH₂^{•+}, m/z 42 (xenon, 80% T// oxygen 80% T).



m/z→

Figure 4. CA mass spectrum of reionized $SiCH_2$ (xenon, 80% T//oxygen 80% T//helium 80% T).



Figure 5. NR mass spectrum of SiCH₃⁺, m/z 43 (xenon, 80% T// oxygen 80% T).

Reduction of SiCH⁺ in NRMS experiments yields an intense recovery signal at m/z 41 (100%). This may reflect the stability of the neutral SiCH[•] and may point to relatively favorable Franck-Condon factors in the vertical electron transfer processes SiCH⁺ \rightarrow SiCH[•] \rightarrow SiCH⁺. The fragmentation pattern is, again, compatible with a connectivity SiCH[•] (1). Further evidence for this assignment is provided by the multistep collision experiment SiCH⁺ (+Xe) \rightarrow SiCH[•] (+O₂) \rightarrow SiCH⁺ (+He) \rightarrow fragment ions. The CA spectrum of reionized SiCH⁺ (Figure 1) thus obtained contains all features characteristic for a connectivity SiCH[•].

The next species studied corresponds to SiCH₂^{•+} (m/z 42), which is also accessible by dissociative ionization of H₃SiCH₂Cl. Its CA mass spectrum is quite simple and it contains fragment ions which are, again, in favor of a connectivity SiCH₂^{•+} (2^{•+}).





Figure 6. CA mass spectrum of reionized SiCH₃ \cdot (xenon, 80% T// oxygen 80% T//helium 80% T).

The base peak is due to loss of $H^{\bullet}(m/z \ 41, 79\%)$; other fragment ions are $m/z \ 40 \ (6\%), \ m/z \ 29 \ (2\%), \ m/z \ 28 \ (13\%), \ m/z \ 14 \ (<1\%).$

Neutralization-reionization of SiCH₂^{•+} (2^{•+}) gives rise to the spectrum shown in Figure 3: The recovery signal at m/z 42 signifies that neutral SiCH₂ (2) is indeed a viable molecule, and the fragmentation pattern is compatible with this assignment. This conclusion is, again, supported by the CA mass spectrum of reionized SiCH₂ (Figure 4).

For the generation of SiCH₃⁺ (m/z 43) H₃SiCH₂Cl turned out to be the most suitable precursor among many other silane derivatives tried. In the CA spectrum of mass-selected m/z 43 we observe the already familiar fragmentation products, i.e. consecutive losses of hydrogen (m/z 42, 36%; m/z 41, 20%; m/z 40, 4%) and formation of m/z 29 (2%), m/z 28 (38%), and of m/z15 (<1%). In view of the thermochemical preference of SiCH₃⁺ (3⁺) versus the remaining isomers (see above), we conclude that it is indeed 3 we are probing experimentally.

Neutralization of 3^+ followed by reionization provides us with a spectrum (Figure 5) which together with the CA spectrum of reionized SiCH₃[•] (Figure 6) permits the conclusion to be drawn that the long sought after SiCH₃[•] is indeed a stable gas-phase molecule.

In conclusion, the present study provides for the first time direct experimental evidence that the neutral molecules $SiCH^{\bullet}$, $SiCH_{2}$, and $SiCH_{3}^{\bullet}$ are viable molecules in the gas phase, thus putting previous conjectures concerning some astrophysical pathways for the genesis of $SiCH_{x}$ fragments on solid experimental ground.

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